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trans-Chloromethyldipyridinepalladium(II)

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Key indicators

Single-crystal X-ray study T = 173 KMean $\sigma(\text{C-C}) = 0.002 \text{ Å}$ R factor = 0.019 wR factor = 0.052Data-to-parameter ratio = 20.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound, $[Pd(CH_3)Cl(C_5H_5N)_2]$, has been synthesized by the reaction of [PdMeCl(COD)] (COD is 1,5-cyclooctadiene) with pyridine in dichloromethane; it is square-planar. The crystal structure features dipole–dipole and π stacking interactions.

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Comment

trans-[Pd(pyridine)₂(Me)Cl], (I), was prepared by addition of an excess of pyridine to [PdMeCl(COD)] (COD is 1,5-cyclooctadiene). Fig. 1 shows the molecular geometry in the crystal structure and the atom-labelling scheme. The crystal structure comprises ordered individual square-planar molecules of trans-[Pd(pyridine)₂(Me)Cl] in a general position; the torsion angles C2-N1-Pd1-C1 and C7-N2-Pd1-C1 are 60.31 (13) and 54.71 (13)°, respectively. The angle between the planes of the pyridine rings is 67.33 (5)°.

Selected geometric parameters are given in Table 1. The bond lengths are in the usual range for $Pd^{II}-C$, Cl, N (Allen *et al.*, 1987). The molecules in the crystal structure are packed in pairs with a $Pd\cdots Pd$ distance of 3.7731 (3) Å. In these pairs, the methyl ligand sits above the chloride ligand and *vice versa* in each case, which may reflect a dipole–dipole interaction between the two molecules (see Fig. 2). Some π stacking [interplanar distance of 3.403 (3) Å] occurs between the

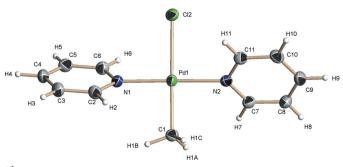


Figure 1
The molecular structure with the atom-labelling scheme. Displacement ellipsoids are shown at the 50% probability level, with H atoms represented as spheres of arbitrary size.

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pyridine rings in adjacent pairs (see Fig. 3), with the rings offset by about one ring width.

The crystal structure is not isostructural with either $[M(\text{pyridine})_2\text{Cl}_2]$ where M = Pd (Viossat *et al.*, 1993) or Pt (Colamarino & Orioli, 1975).

Experimental

A round-bottomed flask was charged with [PdMeCl(COD)] (0.100 g, 0.378 mmol) and CH₂Cl₂ (20 ml). Pyridine (0.15 ml, 1.833 mmol) was added to the solution and the mixture was stirred for 1 h. Hexane (50 ml) was added to the mixture and the volume was reduced to *ca* 20 ml. The resulting white solid was isolated by filtration, washed with two portions of diethyl ether (2 × 20 ml) and dried under vacuum to give a white solid (1.030 g, 0.327 mmol, 87%). A pale-yellow crystal of irregular shape was selected. IR (cm⁻¹, powder film): 1603 (*s*, pyridine). ¹H NMR (CDCl₃): δ 8.80 (*m*, 4H, *o*-pyridine), 7.69 (*m*, 2H, *p*-pyridine), 7.27 (*m*, 4H, *m*-pyridine), 0.73 (*s*, 3H, Pd—CH₃).

Crystal data

[Pd(CH3)Cl(C5H5N)2]	$D_x = 1.727 \text{ Mg m}^{-3}$
$M_r = 315.08$	Mo $K\alpha$ radiation
Monoclinic, C2/c	Cell parameters from 157
a = 13.2867 (9) Å	reflections
b = 11.9185 (8) Å	$\theta = 2.4-27.5^{\circ}$
c = 16.2352 (10) Å	$\mu = 1.72 \text{ mm}^{-1}$
$\beta = 109.5030 \ (10)^{\circ}$	T = 173 (2) K
$V = 2423.5 (3) \text{ Å}^3$	Irregular block, pale yellov
Z = 8	$0.40 \times 0.30 \times 0.20 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan ($SADABS$; Sheldrick, 2003) $T_{min} = 0.583$, $T_{max} = 0.710$	2785 independent reflections 2555 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.030$ $\theta_{\rm max} = 27.5^{\circ}$ $h = -17 \rightarrow 16$ $k = -15 \rightarrow 15$
$T_{\min} = 0.583$, $T_{\max} = 0.710$ 12619 measured reflections	$k = -15 \to 15$ $l = -19 \to 21$

Refinement

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Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.019$	$w = 1/[\sigma^2(F_0^2) + (0.0245P)^2]$
$wR(F^2) = 0.052$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.29	$(\Delta/\sigma)_{\text{max}} = 0.001$
2785 reflections	$\Delta \rho_{\text{max}} = 0.32 \text{ e Å}^{-3}$
137 parameters	$\Delta \rho_{\min} = -0.74 \text{ e Å}^{-3}$

Table 1 Selected geometric parameters (Å, °).

Pd1-N1	2.0441 (13)	Pd1-N2	2.0484 (13)
Pd1-C1	2.0457 (17)	Pd1-Cl2	2.4612 (4)
N1-Pd1-C1	90.58 (6)	N1-Pd1-Cl2	90.15 (4)
N1-Pd1-N2	177.92 (5)	C1-Pd1-Cl2	178.55 (5)
C1-Pd1-N2	88.24 (6)	N2-Pd1-Cl2	90.99 (4)
Cl2-Pd1-N1-C6	56.97 (12)	C1-Pd1-N1-C2	60.31 (13)

H atoms were treated as riding, with C—H distances of 0.95 and 0.98Å and with $U_{\rm iso}({\rm H})$ values of 1.2 and 1.5 times $U_{\rm eq}({\rm C})$ for aromatic and methyl H atoms, respectively.

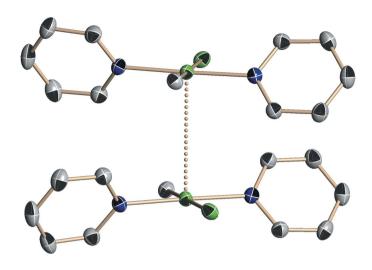


Figure 2
Pair of molecules within the crystal structure. H atoms have been omitted for clarity.

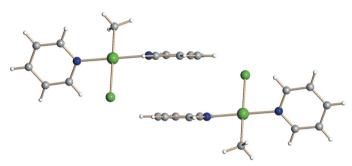


Figure 3 The π stacking in the crystal structure.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

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